



# Gatorbulin-1, a distinct cyclodepsipeptide chemotype, targets a seventh tubulin pharmacological site

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**Tubulin-targeted chemotherapy has proven to be a successful and wide spectrum strategy against solid and liquid malignancies. Therefore, new ways to modulate this essential protein could lead to new antitumoral pharmacological approaches. Currently known tubulin agents bind to six distinct sites at  $\alpha/\beta$ -tubulin either promoting microtubule stabilization or depolymerization. We have discovered a seventh binding site at the tubulin intradimer interface where a novel microtubule-destabilizing cyclodepsipeptide, termed gatorbulin-1 (GB1), binds. GB1 has a unique chemotype produced by a marine cyanobacterium. We have elucidated this dual, chemical and mechanistic, novelty through multidimensional characterization, starting with bioactivity-guided natural product isolation and multinuclei NMR-based structure determination, revealing the modified pentapeptide with a functionally critical hydroxamate group; and validation by total synthesis. We have investigated the pharmacology using isogenic cancer cell screening, cellular profiling, and complementary phenotypic assays, and unveiled the underlying molecular mechanism by in vitro biochemical studies and high-resolution structural determination of the  $\alpha/\beta$ -tubulin–GB1 complex.**

marine natural product | cyanobacteria | total synthesis | microtubules | tubulin

Microtubules are polarized polymers consisting of  $\alpha/\beta$ -tubulin heterodimers involved in cellular structure, motility, proliferation, and intracellular trafficking (1). Pharmacological targeting of tubulin dynamics at different sites (Fig. 1A) has been a validated strategy for cancer therapy for decades and has mostly been linked to the antimitotic effects of these compounds, although increasing evidence has emerged for the importance of nonmitotic effects (1). Natural products targeting tubulin, in particular, have yielded a wealth of chemically diverse agents and provided the basis for several Food and Drug Administration-approved drugs, for both cancer and other pathologies, either alone or as antibody–drug conjugate (ADC), including paclitaxel, vincristine, maytansine, eribulin, and colchicine (Fig. 1B and C). Compounds can be classified based on their binding to one of the six known binding sites, and, even though they are all targeting tubulin, they have shown distinct pharmacological effects. Therefore, there is a persistent interest in the identification of novel microtubule-targeting agents. Two  $\alpha/\beta$ -tubulin binding sites are associated with microtubule stabilization (taxane and laulimalide/peloruside sites; Fig. 1A and B), while binding to four other sites causes microtubule destabilization (vinca, maytansine, colchicine, and pironetin sites; Fig. 1A and C) (1).

Our investigation of marine cyanobacteria as a source of potential anticancer agents has previously yielded the modified peptides dolastatin 10 (Fig. 1C) and dolastatin 15 (2–4), targeting the vinca site (5, 6). Three ADCs with a dolastatin 10 analog (monomethyl auristatin E) as the cytotoxic payload are approved for the treatment of various lymphomas and refractory bladder cancer, while dolastatin 15-based ADCs have advanced to

clinical trials (4). We identified both dolastatins 10 and 15 as indirect hypoxia-inducible factor (HIF) inhibitors based on differential cytotoxicity against a panel of isogenic HCT116 colorectal cancer cells (4, 7), which indicated that HIF inhibition is functionally relevant for the mechanisms of action of these compounds. HIF is activated in solid tumors and promotes metastasis, and targeted screening early in the drug discovery process

## Significance

Natural products provide the inspiration for most drugs, and marine natural products, in particular, are emerging as promising new therapeutics with new targets or mechanisms of action. Pharmacological targeting of tubulin dynamics has been a validated strategy for cancer therapy for decades, yielding structurally diverse natural products and derivatives, including paclitaxel, vincristine, maytansine, and eribulin, targeting six known and different binding sites. We discovered a chemical scaffold from marine cyanobacteria that targets a seventh tubulin binding site. We report the entire spectrum of the discovered chemical and biological novelties, including the isolation, structure determination, and chemical synthesis of the natural product, and the investigation of its mechanism of action, target identification, and binding mode elucidation at the atomic level.

Author contributions: Q.-Y.C., R.R., J.F.D., A.L.R., V.J.P., M.Á.O., and H.L. designed research; S.M., Q.-Y.C., R.R., C.S.F., D.L.-A., F.B., S.T.L., and V.J.P. performed research; A.E.P. contributed new reagents/analytic tools; S.M., Q.-Y.C., R.R., X.W., J.F.D., A.L.R., M.Á.O., and H.L. analyzed data; H.L. wrote the paper; S.M., R.R., X.W., J.F.D., M.Á.O., and H.L. provided data interpretation; S.M. performed isolation, structure determination, and preliminary bioactivity studies; Q.-Y.C. performed total synthesis and analytical compound characterization; R.R. performed mechanistic studies, cellular profiling, cell cycle, COMPARE and gene expression analysis, preliminary tubulin polymerization assays, apoptosis assays, and final figure preparation; C.S.F. performed immunofluorescence, cell cycle, tubulin polymerization assays, and triple-negative breast cancer and drug-resistant panel profiling; D.L.-A. performed ligand displacement assays; F.B. performed crystallization experiments and data collection; A.E.P. provided assistance and troubleshooting with the production and expression of DarPin and the design of the TD1-GA-1 crystallization system; S.T.L. performed angiogenesis, migration, and invasion assays; X.W. supervised angiogenesis, migration, and invasion assays; J.F.D. supervised ligand displacement assays; A.L.R. supervised immunofluorescence, HeLa cell cycle, tubulin polymerization assays, triple-negative breast cancer and drug resistant panel profiling, and data interpretation; V.J.P. collected and identified cyanobacterium and provided cyanobacterial extracts; M.Á.O. supervised crystallization, binding site analysis, and data interpretation and discussion; and H.L. provided overall project supervision, study design and integration, data interpretation, and manuscript preparation.

Competing interest statement: The University of Florida has filed a patent application relating to the content of this article.

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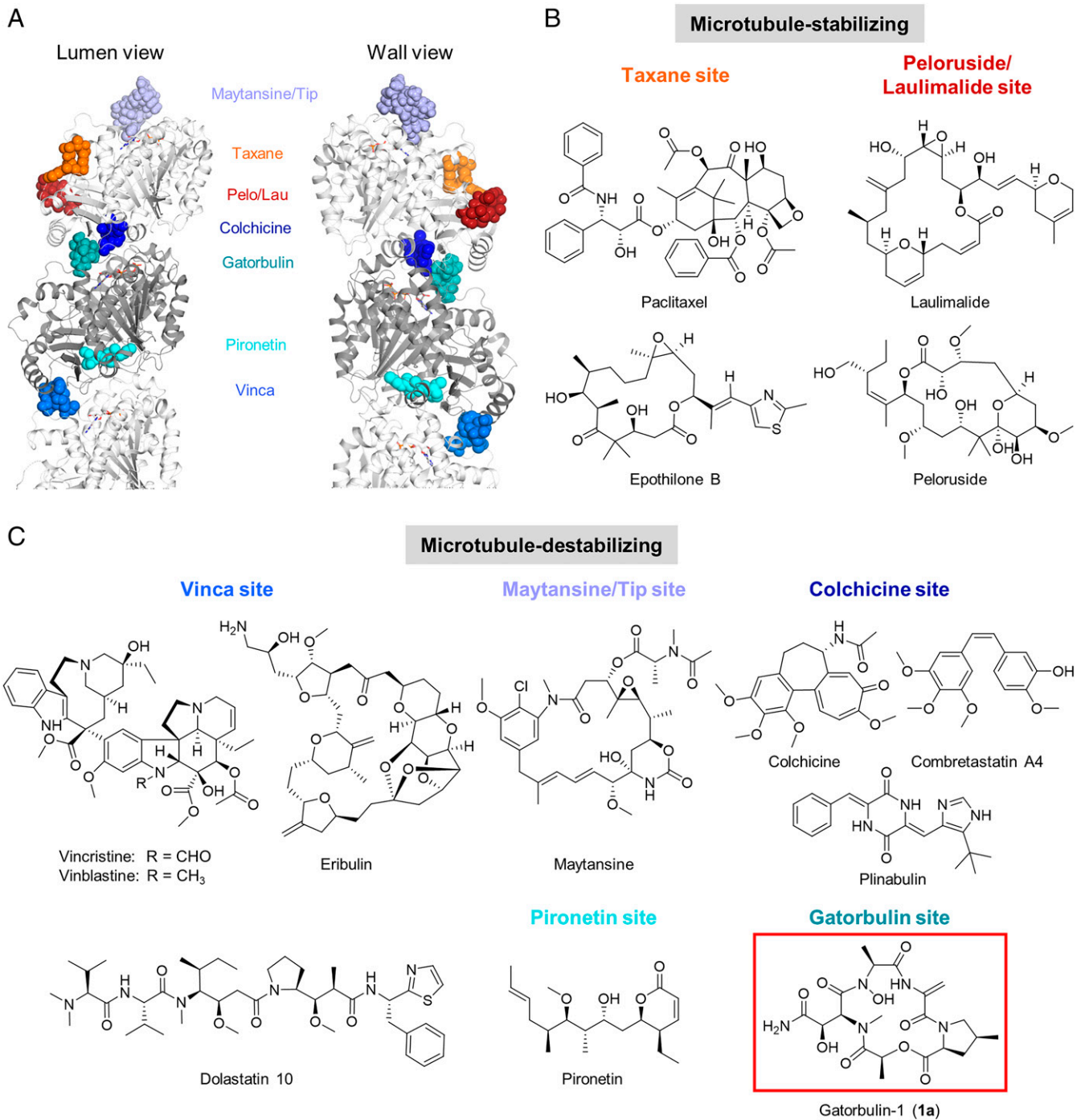
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**Fig. 1.** Binding sites and structures of microtubule-targeting agents. (A) Tubulin heterodimer ( $\alpha$ -tubulin in gray and  $\beta$ -tubulin in white) in ribbon representation, where six known binding sites have been highlighted showing representative ligands in sphere representation: maytansine (PDB ID code 4tv8, violet); epothilone (PDB ID code 4o4i, orange); peloruside (PDB ID code 4o4j, red); colchicine (PDB ID code 4o2b, dark blue); pironetin (PDB ID code 5fnv, cyan), and vinblastine (PDB ID code 4eb6, light blue). The gatorbulin binding site has been also included (PDB ID code 7alr, teal). (B and C) Representative compounds targeting tubulin binding sites. (B) Microtubule-stabilizing agents. (C) Microtubule-destabilizing agents, including the structure of GB1 (1a).

could provide a rapid indication for requisite selectivity for cancer treatment (8–10). Using the same isogenic screening system, we now identified an antiproliferative agent that also possessed preferential activity for oncogenic KRAS-containing and HIF-1 $\alpha$ -containing HCT116 cells and is a microtubule-destabilizing cyclodepsipeptide. We named the compound gatorbulin-1 (GB1, **1a**; Fig. 1C), in analogy to eribulin (Eisai Research Institute), to symbolically represent the discovery of its unique chemical structure

and pharmacological potential at the University of Florida and global Gator Nation partners. Here we report the bioassay-guided isolation, structure determination, synthesis, preliminary structure–activity relationships, mechanism of action, target identification, and binding mode elucidation. Our studies revealed that GB1 represents a unique chemical scaffold targeting a different binding site near the colchicine site and displays distinct pharmacology (Fig. 14).

## Results

**Isolation and Structure Determination of GB1.** Various collections of the cyanobacterium *Lyngbya cf. confervoides* during blooms off the coast near Ft. Lauderdale (11) (Broward County) were extracted with EtOAc–MeOH (1:1). The extracts, previously proven to be rich in secondary metabolites and possessing antifeedant activity (12, 13), were either solvent–solvent partitioned first or directly applied onto a Diaion HP-20 column, and then fractions were subjected to reversed-phase high-pressure liquid chromatography (HPLC) to afford GB1 (**1a**) as an optically active, colorless amorphous solid ( $[\alpha]_D^{20}$  –84.0 [*c* 0.10, MeOH]) and as the most antiproliferative extract component by bioassay-guided isolation using colon cancer cells, along with a minor analog, termed gatorbulin-2 (GB2, **1b**; Fig. 2A). Doubling of virtually every signal in the  $^1\text{H}$  NMR spectrum of **1a** recorded in DMF-*d*<sub>7</sub> suggested the presence of an asymmetric dimer or the presence of conformers in a ratio of 1:1. This observation coupled with the  $[\text{M} + \text{H}]^+$  ion peak at *m/z* 484.2043 obtained by simultaneous electrospray ionization and atmospheric pressure chemical ionization high-resolution mass spectrometry (HRMS) and  $^{13}\text{C}$  NMR data suggested a molecular formula of  $\text{C}_{20}\text{H}_{29}\text{N}_5\text{O}_9$  (calculated for  $\text{C}_{20}\text{H}_{30}\text{N}_5\text{O}_9$ , 484.2044) and consequently the presence of conformers in the NMR solvent. NMR analysis using  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR,  $^1\text{H}$ – $^1\text{H}$  correlation spectroscopy ( $^1\text{H}$ – $^1\text{H}$  COSY), heteronuclear multiple quantum coherence (HMQC), and  $^1\text{H}$ – $^{13}\text{C}$  heteronuclear multiple bond correlation (HMBC) data were carried out for both conformers, revealing two sets of five spin-coupled systems as part of a pentapeptide structure; one signal set appeared slightly broader (Table 1, Fig. 2B, and *SI Appendix*, Figs. S1–S7).

For both signal sets, one putative NH singlet each ( $\delta_{\text{H}}$  8.28 and 8.60) showed COSY correlations to  $\text{sp}^2$ -methylene protons ( $\delta_{\text{H}}$  6.46/5.22 for conformer 1, 6.22/5.11 for conformer 2), which also appeared as singlets in the  $^1\text{H}$  NMR spectrum. Correlations from the NH to the corresponding olefinic methylene carbon ( $\delta_{\text{C}}$  101.8 and 103.0), to a quaternary  $\text{sp}^2$  hybridized carbon ( $\delta_{\text{C}}$  136.3 and 136.2), and to two carbonyl carbons ( $\delta_{\text{C}}$  165.9/170.5 for conformer 1 and 166.8/170.2 for conformer 2) defined the first residue (unit A; Fig. 2A) as a dehydroalanine (DhAla) residue (Table 1 and Fig. 2B).

The second multiproton spin system consisted of two methylenes ( $\delta_{\text{H}}$  4.31/3.25, 2.58/1.58 for conformer 1), two methines ( $\delta_{\text{H}}$  4.42, 2.53), and one methyl group ( $\delta_{\text{H}}$  1.12 d). COSY analysis established their arrangement supported by HMBC data (Table 1 and Fig. 2B). The terminal methine and methylene carbons of this spin system appeared to be nitrogenated ( $\delta_{\text{C}}$  62.6, 56.9), and the HMBC correlation of one of the methylene protons ( $\delta_{\text{H}}$  4.31) to one of the methine carbons ( $\delta_{\text{C}}$  62.6) clarified that the carbons were joined in a 3-methyl pyrrolidine structure (Fig. 2B), which, upon further analysis of HMBC correlations to a carbonyl carbon ( $\delta_{\text{H}}$  170.8), identified the second residue (unit B; Fig. 2A) as a 4-methylproline unit (4-MePro). In a similar fashion, a signal set corresponding to the second conformer for this unit was unambiguously identified (Table 1).

Analysis of the third spin system was straightforward, and this unit consisted of only one methine and a methyl group and NMR data (Table 1 and Fig. 2B), consistent with an acylated lactic acid (Lac) moiety (unit C; Fig. 2A).

Two singlets of another signal set for heteroatom-bound protons ( $\delta_{\text{H}}$  7.26/7.09 for conformer 1 and  $\delta_{\text{H}}$  7.39/7.21 for conformer 2) showed cross-peaks in the COSY spectrum, suggesting a primary amide for the fourth moiety. Another set of singlets at  $\delta_{\text{H}}$  3.09 and 3.14 was indicative of an *N*-methyl tertiary amide group; expectedly, these signals exhibited HMBC correlations with a carbonyl carbon of the adjacent residue ( $\delta_{\text{C}}$  169.9 and 169.7) and, for conformer 1, also to the  $\alpha$ -carbon of the *N*-methylated amino acid ( $\delta_{\text{C}}$  58.0). Rigorous two-dimensional (2D) NMR analysis (Table 1 and Fig. 2B) established the fourth residue (unit D;

Fig. 2A) as an *N*( $\alpha$ )-methyl- $\beta$ -hydroxy-asparagine (*N*( $\alpha$ )-Me- $\beta$ -OH-Asn). Even though the significant broadening of all signals for this unit for the second conformer resulted in fewer HMBC correlations, all  $^1\text{H}$  and  $^{13}\text{C}$  NMR resonances could be assigned except for the  $\alpha$ -carbon, since its NMR signal was too broad to be observed (Table 1).

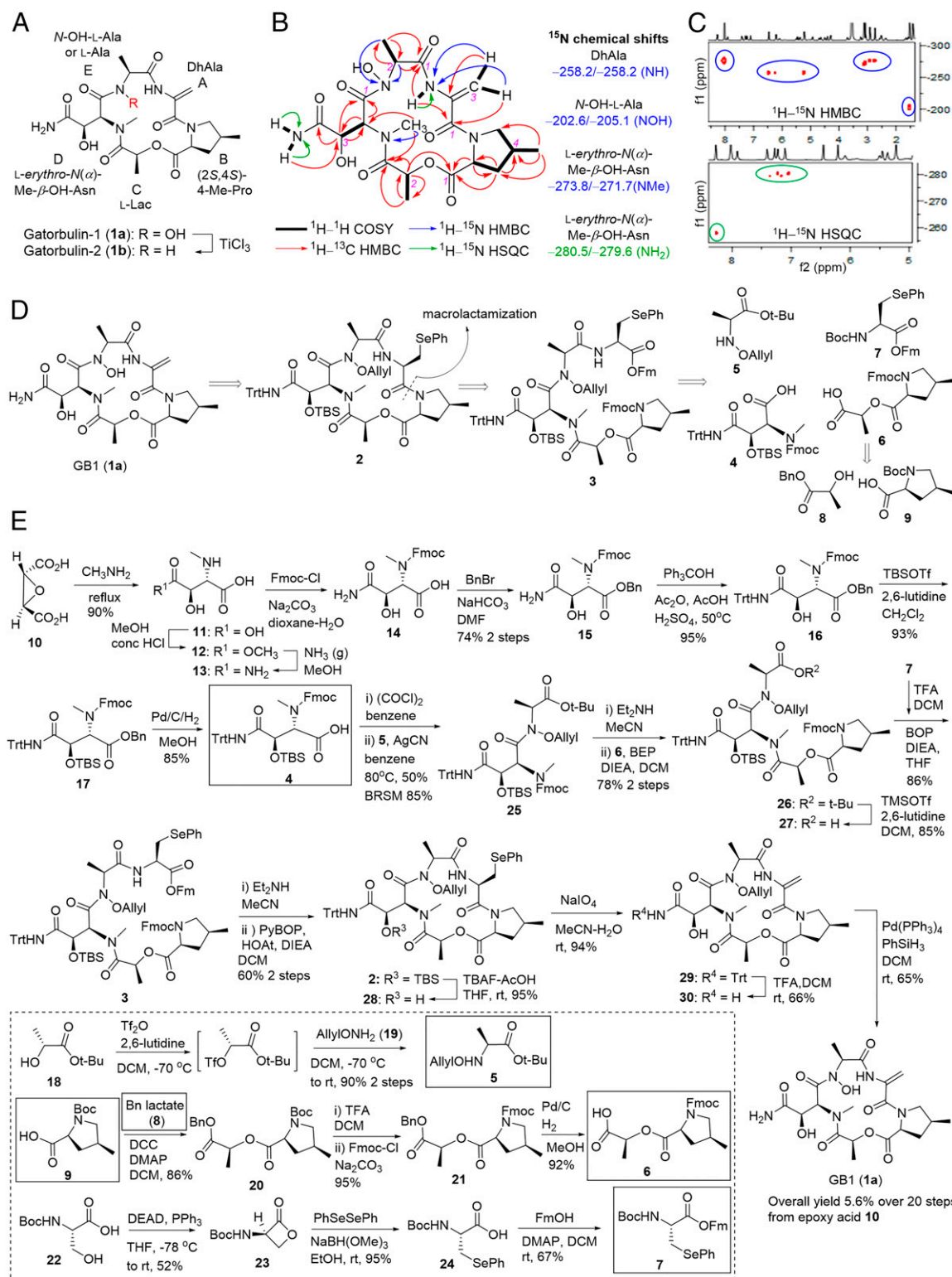
The fifth and last moiety (unit E; Fig. 2A) exhibited similarity to a lactic acid or alanine moiety, yet the  $\alpha$ -carbon resonated at higher field than the corresponding carbon for lactic acid ( $\delta_{\text{C}}$  64.7 and 60.0) and thus more likely bore a nitrogen atom, which then, in turn, had to bear a substituent that was not accounted for yet. This NMR analysis so far led to the assignment of all atoms except one oxygen and hydrogen based on HRMS analysis. In the  $^1\text{H}$  NMR spectrum, the only unassigned signal at this point was a signal for an exchangeable proton at  $\delta_{\text{H}}$  11.35 (br s) for conformer 1 and at  $\delta_{\text{H}}$  10.58 (br) putatively for conformer 2, which could not be rationalized by a secondary amide since it did not show a COSY correlation to the nitrogen-bearing methine and also resonated too far downfield. The chemical shifts were consistent with carboxylic acid protons which could exist in a linear structure; however, it would not leave a substituent for the nitrogen atom in the alanine-like moiety. Therefore, a bond between two heteroatoms, nitrogen and oxygen, had to be invoked, which led us to propose an *N*-hydroxy group in a cyclic hydroxamate (*N*-OH-Ala); its hydroxy proton was also expected to resonate between  $\delta_{\text{H}}$  10 and 12 as observed (Table 1).

The doubling and overlap of signals for several carbonyl carbons for different conformers slightly complicated the sequencing of the individual units. To ultimately prove the existence of the hydroxamate and to validate the nature of the nitrogen atoms, we carried out a  $^1\text{H}$ – $^{15}\text{N}$  HMBC and  $^1\text{H}$ – $^{15}\text{N}$  heteronuclear single quantum correlation (HSQC) analysis (Table 1 and *SI Appendix*, Figs. S8 and S9). HMBC correlations of the H-3a/b methylene protons of the DhAla unit to a nitrogen atom resonating at  $\delta_{\text{N}}$  –258.2 (relative to external  $\text{MeNO}_2$ ,  $\delta_{\text{N}}$  0.0) supported the earlier assignment of a secondary amide (Fig. 2B and C). The *N*-Me protons of the *N*-Me-3-OH-Asn showed HMBC correlations to a nitrogen possessing a chemical shift of  $\delta_{\text{N}}$  –273.8, an expected value for a tertiary amide (Fig. 2B and C). Most importantly and confirmatory for the hydroxamate moiety were two- and three-bond correlations from the  $\alpha$ -methine and the  $\beta$ -methyl protons to a signal at  $\delta_{\text{N}}$  –202.6 (Fig. 2B and C), which is in agreement with literature values for hydroxamate nitrogens [ $\delta_{\text{N}}$  –199.2 for polyoxypeptin A (14)], further validating the proposed structure for GB1 (**1a**). The  $^1\text{H}$ – $^{15}\text{N}$  HSQC spectrum, in addition to the secondary amide proton for DhAla ( $\delta_{\text{N}}$  –258.2), also showed one-bond correlations for both protons of the primary amide functionality for both conformers ( $\delta_{\text{N}}$  –280.5 and –279.6; Fig. 2C). The presence of the hydroxamate functionality was supported analytically by ferric hydroxamate complex formation (15) (*SI Appendix*).

Further support for the proposed structure was found with the isolation of its *N*-deoxy derivative, GB2, or *N*-deoxy-GB1 (**1b**; Fig. 2A). The  $^1\text{H}$  NMR spectrum was strikingly similar to the one of **1a**, including the presence of conformers in a 1:1 ratio in DMF-*d*<sub>7</sub> (*SI Appendix*, Table S1 and Figs. S10–S15). The most significant difference appeared to be the lack of the *N*-OH protons at  $\delta_{\text{H}}$  10 to 12; instead, a new set of doublets appeared in the range for amide protons ( $\delta_{\text{H}}$  8.21 and 8.38 for conformers 1 and 2).

To establish the absolute configuration, we performed acid hydrolysis to liberate the individual units and synthesized all isomers of the amino acid standards for comparative chiral HPLC analysis and advanced Marfey's analysis. The 4-MePro standards were prepared as described previously (16), and the *N*( $\alpha$ )-Me- $\beta$ -OH-aspartic acid stereoisomers were prepared as described in *SI Appendix*. We detected *L*-erythro-*N*( $\alpha$ )-Me- $\beta$ -OH-Asp, (2*S*,4*S*)-4-Me-Pro, and *L*-Lac in the hydrolyzate of **1a**. The structures and identical absolute configurations were confirmed by conversion of





**Fig. 2.** Structure determination and total synthesis of GB1. (A) Structures of the isolated natural products, GB1 (**1a**) and its *N*-deoxy derivative, GB2 (**1b**). GB1 units are labeled A–E. (B) Homonuclear and heteronuclear 2D NMR correlations for GB1. (C) Selected regions of the <sup>1</sup>H–<sup>15</sup>N HMBC and <sup>1</sup>H–<sup>15</sup>N HSQC spectra of GB1. (D) Retrosynthetic analysis for the total synthesis of GB1 (**1a**). (E) Forward synthetic route. Longest linear sequence is shown in the main scheme. Building blocks **4** to **9** from the retrosynthesis (in D) are indicated by boxes with solid lines. The synthesis of the building blocks **5** to **7** is provided in the box with the dashed line.

**1a** into **1b** via TiCl<sub>3</sub>-mediated reduction (Fig. 24) (**17**). Upon acid hydrolysis, **1b** yielded L-Ala as detected by chiral HPLC analysis, establishing the remaining stereogenic center.

**Structure Validation by the Total Synthesis of GB1.** To prove the structure and overcome the supply issue, we embarked on total synthesis. The retrosynthetic analysis of total synthesis of GB1 (**1a**)

Table 1. NMR data for GB1 (1a) for both conformers (1:1) in DMF-*d*<sub>7</sub>

| Unit   | C/H no.                 | Conformer 1              |                                |                         |  | Conformer 2                           |   |                          |                                |                         |
|--|-------------------------|--------------------------|--------------------------------|-------------------------|--|---------------------------------------|---|--------------------------|--------------------------------|-------------------------|
|  |                         | $\delta_H$ (J in Hz)*    | $\delta_C$ , mult <sup>†</sup> | $\delta_N$ <sup>‡</sup> | COSY*  | <sup>1</sup> H- <sup>13</sup> C HMBC* | <sup>1</sup> H- <sup>15</sup> N HMBC <sup>§</sup> | $\delta_H$ (J in Hz)*    | $\delta_C$ , mult <sup>†</sup> | $\delta_N$ <sup>‡</sup> |
| DhAla<br>(A)                                     | 1                       |                          | 165.9, qC                      |                         |  |                                       |   |                          | 166.8, qC                      |                         |
|  | 2                       |                          | 136.3, qC                      |                         |  |                                       |   |                          | 136.2, qC                      |                         |
|  | 3a                      | 6.46, s                  | 101.8, CH <sub>2</sub>         |                         | H-3b   | 1, 2                                  |   | 6.22, s                  | 103.0, CH <sub>2</sub>         |                         |
|  | 3b                      | 5.22, s                  |                                |                         | H-3 <sup>a</sup>                             | 1                                     |   | 5.11, s                  |                                |                         |
|  | NH                      | 8.28, s                  |                                | -258.2                  |  | 1, 1 (Ala)                            | H-3a, H-3b  | 8.60, s                  |                                | -258.2                  |
| 4-Me-Pro<br>(B)                                  | 1                       |                          | 170.8, qC                      |                         |  |                                       |   |                          | 170.4, qC                      |                         |
|  | 2                       | 4.42, t (7.0)            | 62.6, CH                       |                         | H-3a, H-3b                                   | 1, 3                                  |   | 4.94, dd (9, 8)          | 63.8, CH                       |                         |
|  | 3a                      | 2.58, m                  | 37.0, CH <sub>2</sub>          |                         | H-2, H-3b, H-4                               | 1, 2, 4, 5, 6                         |   | 2.53, m                  | 40.8, CH <sub>2</sub>          |                         |
|  | ( <i>proS</i> )         |                          |                                |                         |  |                                       |   |                          |                                |                         |
|  | 3b                      | 1.58, m                  |                                |                         | H-2, H-3a, H-4                               | 1, 2, 4, 5, 6                         |   | 1.65, m                  |                                |                         |
|  | ( <i>proR</i> )         |                          |                                |                         |  |                                       |   |                          |                                |                         |
|  | 4                       | 2.53, m                  | 34.7, CH                       |                         | H-3a, H-3b, H-5a, H-5b,<br>H <sub>3</sub> -6 | 3, 5, 6                               |   | 2.38, m                  | 32.4, CH                       |                         |
|  | 5a                      | 4.31, dd (-10.0,<br>6.0) | 56.9, CH <sub>2</sub>          |                         | H-4, H-5b                                    | 2, 3, 4, 6                            |   | 3.71, dd (-11.0,<br>7.0) | 55.6, CH <sub>2</sub>          |                         |
|  | ( <i>proR</i> )         |                          |                                |                         |  |                                       |   |                          |                                |                         |
|  | 5b                      | 3.25, dd (-10.0,<br>8.5) |                                |                         | H-4, H-5 <sup>a</sup>                        | 3, 6                                  |   | 3.11, dd (-11.0,<br>11)  |                                |                         |
| 6  | 1.12, d (7.0)           | 18.0, CH <sub>3</sub>    |                                | H-4                     | 3, 4, 5                                      |                                       | 1.11, d (7.0)                                     | 16.9, CH <sub>3</sub>    |                                |                         |
| Lac<br>(C)                                       | 1                       |                          | 169.9, qC                      |                         |  |                                       |   |                          | 169.7, qC                      |                         |
|  | 2                       | 5.47, q (7.0)            | 72.9, CH                       |                         | H <sub>3</sub> -3                            | 1, 3, 1<br>(MePro)                    |   | 5.39, q (7.0)            | 72.1, CH                       |                         |
| <i>N</i> ( $\alpha$ )-Me- $\beta$ -OH-Asn<br>(D) | 3                       | 1.42, d (7.0)            | 17.6, CH <sub>3</sub>          |                         | H-2  | 1, 2                                  |   | 1.41, d (7.0)            | 17.3, CH <sub>3</sub>          |                         |
|  | 1                       |                          | 170.4, qC                      |                         |  |                                       |   |                          | 169.4, qC                      |                         |
|  | 2                       | 5.51, br d (9.5)         | 58.0 (br),<br>CH               |                         | H-3,   | 1, 2                                  |   | 5.85                     | Not<br>observed                |                         |
|  | 3                       | 4.51, dd (9.5,<br>4.5)   | 68.8, CH                       |                         | H-2, 3-OH                                    | 1, 2, 4                               |   | 4.65, br dd (6, 6)       | 69.7, CH                       |                         |
|  | 3-OH                    | 6.05, br d               |                                |                         | H-3  |                                       |   | 5.06, br                 |                                |                         |
|  | 4                       |                          | 175.1, qC                      |                         |  |                                       |   |                          | 174.8                          |                         |
|  | <i>N</i> -Me            | 3.09, s                  | 34.0, CH <sub>3</sub>          | -273.8                  |  | 2, 1 (Lac)                            | <i>N</i> -Me                                      | 3.14, s                  | 34.5 (br),<br>CH <sub>3</sub>  | -271.7                  |
|  | 4-NHa                   | 7.26, s                  |                                | -280.5                  | 4-NHb  |                                       |   | 7.39, s                  |                                | -279.6                  |
|  | 4-NHb                   | 7.09, s                  |                                |                         | 4-NHa  | 3                                     |   | 7.21, s                  |                                |                         |
|  | <i>N</i> -OH-Ala<br>(E) | 1                        |                                | 170.5, qC               |  |                                       |   |                          |                                | 170.2, qC               |
| 2  |                         | 4.29, q (6)              | 64.7, CH                       |                         | H <sub>3</sub> -3, NH                        | 1, 3                                  |   | 4.71, br q (6)           | 60.0, CH                       |                         |
| 3  |                         | 1.54, d (6)              | 13.8, CH <sub>3</sub>          |                         | H-2  | 1, 2                                  |   | 1.42, d (6)              | 15.6, CH <sub>3</sub>          |                         |
| <i>N</i> -OH                                     |                         | 11.35, br s              |                                | -202.6                  |  |                                       | H-2, H <sub>3</sub> -3                            | 10.58, br                |                                | -205.1                  |

br: broad, s: singlet, d: doublet, q: quartet, dd: doublet of doublets, qC: quaternary carbon.

\*Recorded at 600 MHz.

<sup>†</sup>Recorded at 150 MHz.

<sup>‡</sup>Deduced from <sup>1</sup>H-<sup>15</sup>N HSQC and <sup>1</sup>H-<sup>15</sup>N HMBC (500 MHz).

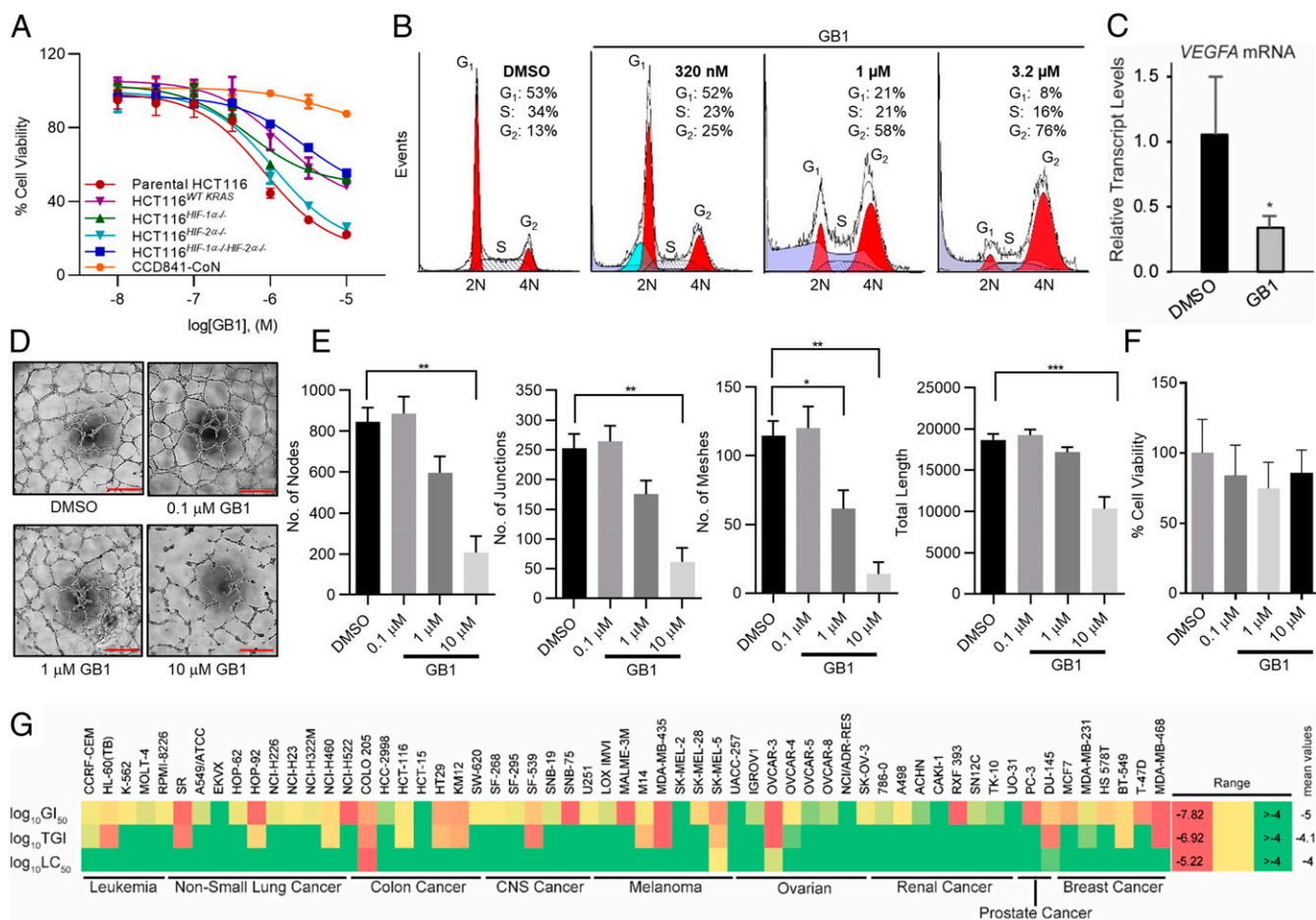
<sup>§</sup>Recorded at 500 MHz.

is shown in Fig. 2D. The final product **1a** could be obtained from the fully masked cyclized precursor **2** by sequential deprotection. The site between 4-MePro and (Se)-phenylselenocysteine [Sec(Ph)] was chosen for macrolactamization. In linear precursor **3**, the 9-fluorenylmethoxycarbonyl (Fmoc) and 9-fluorenylmethyl (Fm) pair was designed as the protection groups of amino and carboxy termini, which could be cleaved simultaneously with base to provide the precursor of macrocyclization (**18**). Sec(Ph) was proposed as the pronit of DhAla (**19**). Linear compound **3** was disconnected into four building blocks **4** to **7**, which could be constructed from commercially available reagents (e.g., **6** from **8** and **9**) using established or modified protocols.

Fig. 2E depicts the synthetic route to GB1 (**1a**). The synthesis of acid **4** was adopted from published procedures (**20**). The (2*R*,3*R*)-epoxysuccinic acid (**10**) was converted to *erythro-N*( $\alpha$ )-methyl-3-hydroxy-L-aspartic acid (**11**) by treatment with methylamine-water

under reflux. Then **11** was selectively esterified with acidic methanol under refluxing (**21**) to provide monoester **12**. Without purification, aminolysis of **12** with ammonia (gas) in MeOH provided *N*( $\alpha$ )-Me- $\beta$ -OH-Asn (**13**) (**20**, **21**), which had poor solubility in MeOH, so that pure product could be obtained by simple filtration. Sequential protections of the groups of **13** using standard methods (**22**) provided full masked compound **17**. Finally, acid **4** was obtained from **17** by hydrogenation with palladium catalyst.

The synthesis of the building block of allyloxamine **5** adopted the triflate method (**23**). Allyl was chosen as the NOH protecting group as it could be selectively removed by Pd(Ph<sub>3</sub>P)<sub>4</sub> in the presence of dehydropeptide. Acid **6** was synthesized from (4*S*)-*N*-Boc-4-methyl-Pro (**9**) and benzyl-L-lactate (**8**) by standard protocols of esterification, protection, and deprotection. Following established procedures (**19**), *N*-Boc-serine (**22**) was converted



**Fig. 3.** Mechanism of action of GB1 through cellular profiling. (A) Antiproliferative activity of GB1 in parental HCT116 colon cells, isogenic HCT116 knockout cells, and CCD-841CoN normal epithelial colon cells (48 h treatment). GB1 showed a marginal effect on the viability of CCD-841CoN normal epithelial colon cells (0.5% DMSO vehicle). Parental HCT116 cells and HCT116<sup>HIF-2α-/-</sup> were most susceptible, while the potency and efficacy of GB1 were reduced against HCT116<sup>HIF-1α-/-HIF-2α-/-</sup>, HCT116<sup>HIF-1α-/-</sup>, and oncogenic KRAS knockout (HCT116<sup>WT KRAS</sup>). Cell viability was measured by MTT assay ( $n = 3$ ). Data are represented as average  $\pm$  SD. (B) Cell cycle analysis. HCT116 cells were treated with GB1 (320 nM, 1  $\mu$ M, or 3.2  $\mu$ M) or vehicle control (0.25% DMSO) for 24 h, and DNA content was assessed by flow cytometry of propidium iodide stained cells. GB1 induced G<sub>2</sub>/M accumulation. (C) HIF target gene (*VEGFA*) expression after 16-h exposure of parental HCT116 cells to GB1 (3.2  $\mu$ M) or vehicle (0.25% DMSO). RNA was isolated, reverse transcribed, and subjected to qPCR using TaqMan analysis.  $\beta$ -actin served as endogenous control. Error bars indicate mean  $\pm$  SD of three replicates (Student *t* test, \* $P < 0.05$ , \*\* $P < 0.01$ , and \*\*\* $P < 0.001$ ). (D–F) GB1 inhibits HUVEC tube formation in vitro without toxicity. (D) Representative images of HUVEC tube formation in growth factor-reduced Matrigel upon treatment with DMSO (0.1%) or varying concentrations (10  $\mu$ M, 1  $\mu$ M, or 0.1  $\mu$ M) of GB1 (9 h). (Scale bar: 200  $\mu$ m.) All images shown are representative, and data are represented as mean  $\pm$  SEM; one-way ANOVA followed by Tukey's multiple comparisons test; \* $P < 0.05$ , \*\* $P < 0.01$ , and \*\*\* $P < 0.001$ . (E) Quantification of number of nodes, number of junctions, number of meshes, and total length of tubes ( $n = 3$ ). (F) GB1 does not affect HUVEC cell viability. HUVEC cell viability was quantified by absorbance at 490 nm using (3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium, inner salt) (MTS assay, 24 h). GB1 treatment did not affect total number of viable HUVEC cells compared to DMSO-treated control ( $n = 3$ ). For E and F, data are represented as mean  $\pm$  SEM; one-way ANOVA followed by Tukey's multiple comparisons test. (G) Heatmap for the performance of GB1 across cell lines in the NCI-60 screen using three different values (growth-inhibitory effect, G<sub>150</sub>; cytostatic effect, TGI; cytotoxic effect, LC<sub>50</sub>; concentration in molar).

to BocSec(Ph) (**24**) via  $\beta$ -lactone **23**, which was esterified with FmOH to provide building block **7**. DhAla could be obtained from phenylselenocysteine by oxidative  $\beta$ -elimination.

The fusion of building blocks was initiated by coupling of acid **4** with allyloxamine **5**. Acid **4** was activated to the acid chloride, which was then coupled with **5** in the presence of AgCN (**24**, **25**) to provide **25** in 50% isolated yield and 85% yield based on recovered starting materials. This acylation was not successful when other common coupling reagents were used, because of poor nucleophilicity of the nitrogen and high steric hindrance (**26**). Compound **26** was obtained in 78% yield by 2-bromo-1-ethyl-pyridinium tetrafluoroborate-mediated coupling (**27**) of acid **6** with the free methylamine generated by selective deprotection of **25**. Selective removal of the *t*-butyl group of **26** by trimethylsilyl trifluoromethanesulfonate/2,6-lutidine (**28**) afforded

acid **27**, which was coupled with free amine from **7** using (benzotriazol-1-yloxy)tris(dimethylamino)phosphonium hexafluorophosphate as a coupling reagent to yield the linear compound **3** in 86% yield. The use of the *t*-butyl protecting group prevented diketopiperazine (**29**) formation upon coupling of **6** and **25**, and ensured that trityl and *tert*-butyldimethylsilyl (TBS) groups were intact for the generation of acid **27**. Both Fmoc and Fm protection groups were removed simultaneously when compound **3** was exposed to Et<sub>2</sub>NH in MeCN. The macrocyclization was mediated by PyBOP/HOAt to give macrocycle **2** in 60% yield for two steps. Removal of TBS of **2** with the tetrabutylammonium fluoride/HOAc buffer (**22**, **30**) provided **28**, and subsequent oxidation of SePh with NaIO<sub>4</sub> (**19**) yielded dehydropetide **29**. Trityl was removed with trifluoroacetic acid in CH<sub>2</sub>Cl<sub>2</sub> (**31**) to yield primary amide **30**. The removal of allyl group by Pd(PPh<sub>3</sub>)<sub>4</sub>/PhSiH<sub>3</sub> (**30**, **32**)



provided final product **1a**. The removal sequences for trityl and allyl groups are interchangeable; however, the yield would drop from 66 to 35%. GB1 (**1a**) was purified by reversed-phase thin-layer chromatography plate (C18). The overall yield over 20 steps from the epoxy acid **10** was 5.6% (see *SI Appendix* for details and *SI Appendix*, Figs. S16–S61 for all NMR spectra). The synthetic sample was identical to the isolated natural product, which was verified by NMR, HRMS, and optical rotation (*SI Appendix*, Figs. S62 and S63).

**GB1 Cellular Profiling Identifies the Mechanism of Action.** GB1 (**1a**) was identified as the extract's active component against colon cancer cells and showed a concentration that inhibits response by 50% ( $IC_{50}$ ) of 0.80  $\mu$ M in a 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-2H-tetrazolium bromide assay against HCT116 colorectal cancer cells (Fig. 3A), while **1b** was inactive at the highest concentration tested ( $IC_{50} > 10 \mu$ M), indicating that the hydroxamate moiety is indispensable to the antiproliferative activity. Isogenic cell line selectivity screening indicated preferential activity against parental HCT116 compared with the oncogenic KRAS knockout or double knockout of both HIF-1 $\alpha$  and HIF-2 $\alpha$  transcription factors (Fig. 3A) (7). Deconvolution using single knockouts of HIF-1 $\alpha$  and HIF-2 $\alpha$  clearly demonstrated that only cells depleted in HIF-1 $\alpha$  had reduced susceptibility to GB1, which is consistent with HIF-1 $\alpha$  being in the same pathway and activated by oncogenic KRAS. Furthermore, normal epithelial colon cells (CCD841-CoN) were less inhibited, indicating an additional promising level of selectivity (Fig. 3A). The preference for HIF-1 $\alpha$ -expressing cells parallels the selectivity profiles for microtubule agents we previously discovered (dolastatins 10 and 15) (4, 7). DNA content analysis revealed the concentration-dependent G<sub>2</sub>/M cell cycle accumulation characteristic for antimetabolic agents (Fig. 3B). A concentration of GB1 ( $4 \times IC_{50}$ ) that results in a complete antiproliferative response initiated a concomitant down-regulation of the HIF-1 $\alpha$  target gene *VEGFA* in parental HCT116 cells (Fig. 3C) that was even more pronounced than for dolastatin 15 (4), which also elevated GB1's potential as antiangiogenic agent. Consistent with the role of microtubule dynamics in angiogenesis (33), GB1 also inhibited endothelial tube formation without cytotoxicity (Fig. 3D–F). Endothelial cell tube formation in Matrigel mimics the process of blood vessel formation in vivo (33). Our study suggested that 10  $\mu$ M GB1 was necessary to block the ability of human umbilical vein endothelial cells (HUVECs) to form tube-like structures within 3 h (Fig. 3D). Quantification of the structural network revealed a significant reduction in the number of nodes, number of junctions, number of meshes, and total length of tubes (Fig. 3E). Importantly, GB1 did not affect HUVEC cell viability, even after 24 h (Fig. 3F). Taken together, this reveals an antiangiogenic potential of the compound in inhibiting the formation and stabilization of a 3D vascular network. The additional *VEGF* down-regulation in growth factor-secreting cells, which is a driver of angiogenesis, is expected to result in a pronounced antiangiogenic action of GB1 in vivo.

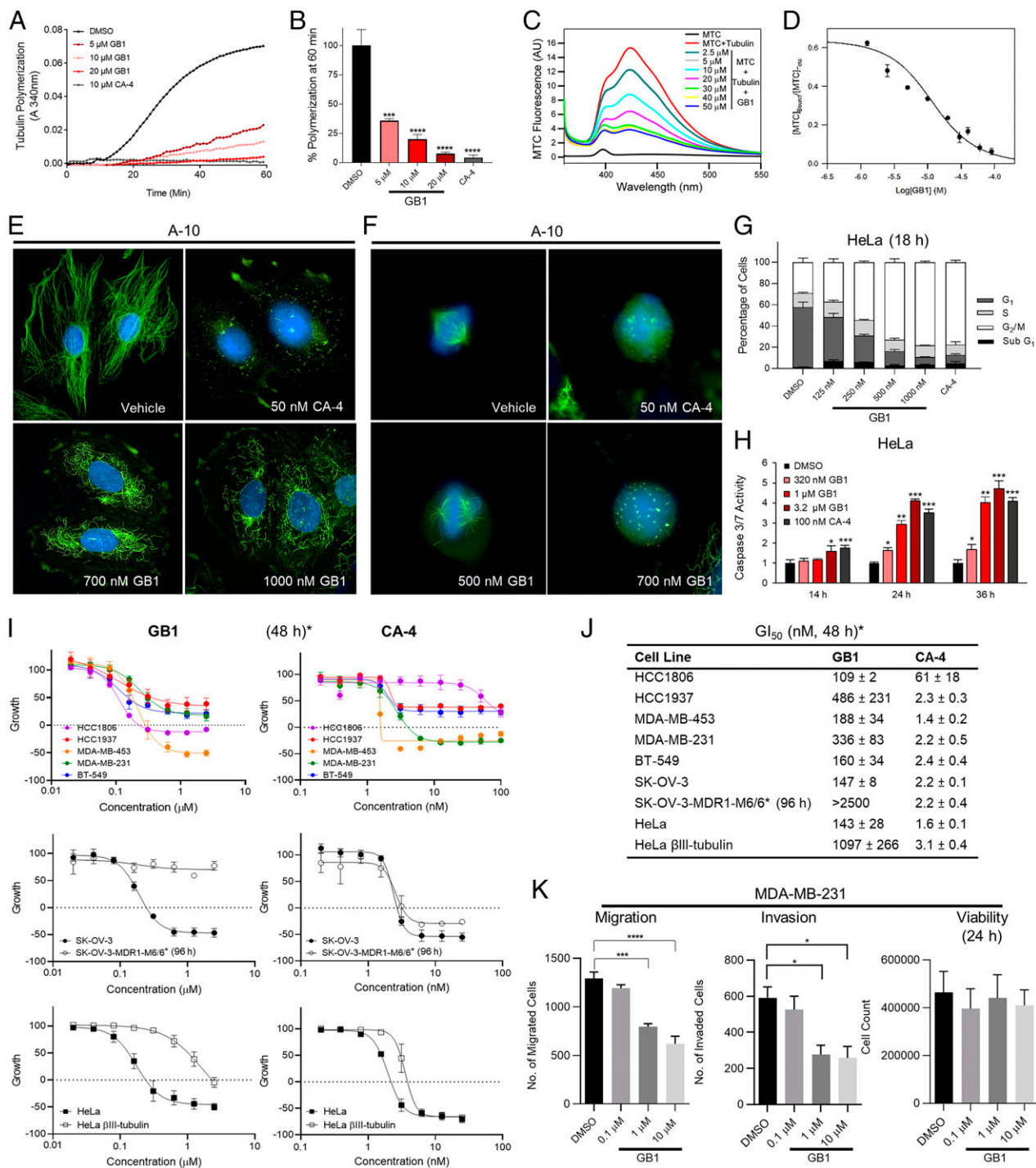
The NCI-60 cell line screen data (Fig. 3G), analyzed by the COMPARE algorithm (34), were indicative of a cytotoxicity profile most related to antimetabolic/tubulin agents, including paclitaxel, eribulin, colchicine, and vinca bis-indole alkaloid derivatives ( $P$  0.75 to 0.85), suggesting that the biochemical mechanisms of action are related. While GB1 displayed an  $IC_{50}$  of  $>10 \mu$ M against normal mucosal colon cells, it had submicromolar activity against HCT116 cells ( $GI_{50}$  306 nM) and was even more potent against COLO205 cells ( $GI_{50}$  92 nM) based on the NCI-60 data (sulforhodamine B assay). Other susceptible cell types included certain melanoma (SK-MEL-5), ovarian (OVCAR-3), and prostate (DU-145) cancer cells, which correspond to certain cancers where microtubule agents have been successful. Furthermore, cervical and breast cancers are relevant indications, prompting additional studies in these cell types (see below) (35).

**Target Identification and GB1 Underlying Molecular Mechanism.** First, using a biochemical assay, we demonstrated that GB1 directly inhibits tubulin polymerization in vitro in an iron-independent manner (Fig. 4A and B). Second, we tested GB1 binding to commonly targeted sites for microtubule-destabilizing agents, including the vinblastine, maytansine, and colchicine sites (Fig. 1A), by employing fluorescent bona fide probes (36–38). GB1 was unable to displace the fluorescent derivatives of eribulin and maytansine, indicating that it did not bind to these sites. However, GB1 successfully competed with a fluorescent probe of the colchicine site [2-methoxy-5-(2,3,4-trimethoxyphenyl)-2,4,6-cycloheptatrien-1-one (MTC)] (Fig. 4C and D) with an apparent binding affinity of  $1.01 \pm 0.08 \times 10^6 M^{-1}$ , which is below colchicine apparent binding affinity [ $1.5 \times 10^7 M^{-1}$  (39)].

We performed subsequent cell-based studies with GB1 in comparison with combretastatin A4 (CA-4), a known colchicine site binder. We demonstrated that GB1 promotes microtubule depolymerization in A-10 (rat smooth muscle) interphase cells, similar to CA-4, although with lower potency (Fig. 4E). GB1 promoted aberrant mitotic spindle formation at submicromolar concentrations (Fig. 4F). HeLa cervical cancer cells were highly susceptible to GB1; the compound induced accumulation of cells in the G<sub>2</sub>/M cell cycle phases after 18 h, with low levels of apoptosis as suggested by an emerging sub-G<sub>1</sub> population (Fig. 4G). We then measured the caspase-3/7 activity in a concentration- and time-dependent manner, validating that treatment with GB1 induces apoptosis at concentrations that affect tubulin dynamics in cells, especially after 24 h and beyond, similar to CA-4 (Fig. 4H). We then compared the antiproliferative effects of GB1 and CA-4 in a panel of triple-negative breast cancer (TNBC) cells and in drug-resistant models, including parental and Pgp-overexpressing SK-OV-3 ovarian adenocarcinoma cells (48 and 96 h of treatment, respectively) and parental and  $\beta$ III-expressing HeLa cells (48 h) (40). The compounds showed a different fingerprint. GB1 does not circumvent Pgp-mediated or  $\beta$ III-tubulin-mediated drug resistance (Fig. 4I), indicating that it was mechanistically distinct from CA-4. Additionally, it has a distinct profile of relative efficacy against a panel of TNBC cell lines as compared to CA-4 (48 h; Fig. 4I and J). MDA-MB-231 cells pretreated with GB1 for 24 h demonstrated a significant concentration-dependent inhibition of cancer cell migration and invasion abilities across a transwell membrane as compared to dimethyl sulfoxide (DMSO)-treated controls, without significantly affecting cell viability (Fig. 4K). This phenotype is also consistent with the microtubule-targeting mechanism (41). Taken together, the mechanistic biochemical and pharmacological data suggested that GB1 can block ligand binding to the colchicine site but may not fit in the classic pocket, resulting in a distinct profile of cellular activity.

To further probe our hypothesis and identify the specific binding mechanism, we aimed at high-resolution macromolecular crystallography. The T<sub>2</sub>R-TTL system (two  $\alpha/\beta$ -tubulin dimers in complex with the stathmin-like protein RB3 and tubulin tyrosine ligase) has successfully produced several structures of ligand bound to tubulin using X-ray crystallography (42). However, GB1 soaking experiments did not reveal any difference density related to the ligand. We finally succeeded using the T<sub>1</sub>-DARPin (T1D) complex that was previously used to solve the structure of colchicine-bound tubulin by serial millisecond crystallography (43). We solved the structure of T1D-GB1 at 1.94-Å resolution (*SI Appendix*, Table S2), and we unequivocally found ligand density at a site that is close to but clearly different from the colchicine binding site (Fig. 5A). This binding site sits at the intradimer interface between  $\alpha$ - and  $\beta$ -tubulins. It is equivalent to the vinca site, but the defined pocket boundaries are reversed (Fig. 5B), meaning that contacts with loop T7, helix H10, and strand S9 are provided by  $\beta$ -tubulin (whereas these are at  $\alpha$ -tubulin in the vinca site) and  $\alpha$ -tubulin supplies interactions by loops T5 and H6–H7 (which correspond to  $\beta$ -tubulin in the vinca site).

The structure superimposes very well with the T1D apo-structure (Protein Data Bank [PDB] ID code 4drx) with an overall rmsd of



**Fig. 4.** Target identification and selectivity profiling of GB1 in comparison with combretastatin A-4 (CA-4). (A and B) Inhibition of tubulin polymerization by GB1. (A) Time course of tubulin polymerization (20 μM) (vehicle: DMSO; 1%). (B) Quantification of tubulin polymerization at 60 min. (C and D) Displacement of MTC. (C) Fluorescence emission spectra of 10 μM MTC in the absence (black line) and the presence (red line) of 10 μM tubulin plus varying concentrations of GB1 (vehicle: DMSO; 1%). (D) Displacement isotherm at 25 °C of MTC by GB1. Line is best fit of the GB1 equilibrium constant, assuming 0.8 sites per tubulin dimer. (E) Microtubule depolymerization GB1 in interphase cells. A-10 cells treated for 18 h (vehicle: DMSO; 0.1%). (Magnification: 1,000x.) (F) GB1 promotes formation of aberrant mitotic spindles in A-10 cells (18 h treatment). (G) GB1 arrest cell cycle in G<sub>2</sub>/M phase. Cell cycle distribution of HeLa cells in the presence of GB1 (vehicle: DMSO; 0.1%) (H) GB1 causes apoptosis in HeLa cells as measured by Caspase-3/7 activity (vehicle: DMSO; 0.5%). (I and J) Antiproliferative and cytotoxic effects of GB1 and CA-4 in a panel of TNBC cell lines, in parental and Pgp-expressing SK-OV-3 cells, or in parental and βIII-expressing HeLa cells (sulforhodamine B assay, vehicle: DMSO; 0.5%). Asterisk denotes that treatment time was 48 h, except for SK-OV-3-MDR1-M6/6\* (96 h). (K) Effects of GB1 on transwell migration and invasion of serum-starved MDA-MB-231 breast cancer cells pretreated with GB1 for 24 h. Pretreated cells were allowed to migrate or invade for 5 h. Quantitative analysis of migrated or invaded cells across a Matrigel layer (vehicle: DMSO; 0.1%). Cell count (24 h) was determined by Trypan Blue exclusion assay. (A, C, E, and F) Representative set from three experiments shown. (B and G–K) Error bars represent ± SEM (B, G, and I–K) or SD (H) from three independent experiments. (B) One-way ANOVA with Dunnett's post hoc test; \*\*\**P* < 0.001, \*\*\*\**P* < 0.0001. (D and K) *n* = 3. (E and F) Microtubules visualized using a β-tubulin antibody (green) and DNA was visualized using DAPI (blue). (H) Two-tailed Student *t* test; \**P* < 0.05, \*\**P* < 0.01, \*\*\**P* < 0.005. (K) One-way ANOVA followed by Tukey's multiple comparisons test; \**P* < 0.05, \*\*\**P* < 0.001, \*\*\*\**P* < 0.0001.





molecule to the  $\beta$ V355 (strand S9) amide nitrogen. The hydroxamate of this alanine (*N*-OH-Ala) hydrogen bonds to  $\beta$ D329 (helix H10) side chain O62 and O61 (Fig. 5C), rationalizing the critical function of the hydroxamate for binding. The hydroxy propanamide residue (*N*( $\alpha$ )-Me- $\beta$ -OH-Asn) makes extensive contacts with  $\alpha$ - and  $\beta$ -tubulins through loop T5, including the carbonyl moiety of  $\alpha$ P175 and  $\alpha$ Q176, the carbonyl and the amide of  $\alpha$ S178, and strand S9, with  $\beta$ K352 main chain and  $\beta$ V353 amide nitrogen (Fig. 5D). Additionally, the *N*-Me group of this residue further hydrogen bonds to  $\beta$ V353 carbonyl moiety. The carbonyl of the Lac that closes the ring with 4-MePro further stabilizes contacts with the  $\alpha$ -tubulin loop T5 through hydrogen bond to the carbonyl of  $\alpha$ Q176.

The tubulin–colchicine (PDB ID code 5nm5) and tubulin–GB1 structures superimpose with an overall rmsd of 0.497 Å over 756  $C_{\alpha}$ -atoms, with the main differences located at  $\beta$ -tubulin loop T7 and  $\alpha$ -tubulin loop T5 (Fig. 5E). These loops imperatively change their conformations upon colchicine binding because, otherwise, the trimethoxy phenyl ring of colchicine (Fig. 1C) would clash with  $\beta$ L247 and  $\beta$ N248 (loop T7) and the amide of ring B would clash with the  $\alpha$ T179 (loop T5). The presence of GB1 at the site precludes the simultaneous binding of a second ligand in the colchicine binding site due to GB1 interactions with those loops, thus correlating with the results from our competition assay in which increasing concentrations of GB1 prevent any binding of the MTC fluorescence probe to tubulin (Fig. 4 C and D).

## Discussion

We took an integrated approach toward natural products drug discovery by targeting minor, highly bioactive compounds from a chemically prolific cyanobacterium, combining innovative screening and rigorous bioassay-guided isolation and structure determination with chemical synthesis to overcome the supply problem, and achieving in depth mechanistic studies as well as direct target and binding site identification. We advocated for such an approach to fully exploit the proven potential of natural products and increase the value of bioactive natural products (44). The ultimate key for a successful natural product drug discovery campaign is the choice of the source organism. We have been focusing on marine cyanobacteria, which are prolific yet underexplored marine prokaryotes with a tremendous biosynthetic potential. The gatorbulin-yielding sample was derived from a blooming “superproducer” of secondary (specialized) metabolites (natural products) that previously yielded linybyastatins 4 to 6, pompanopeptins A and B, tiglicamides, and larginides/larginide D oxazolidine, most of which are non-cytotoxic serine protease inhibitors (12, 13, 45–47). Beyond showcasing the biosynthetic capacity of marine cyanobacteria, our discovery of GB1 exemplifies that marine cyanobacterial natural products occupy therapeutically relevant chemical space that could lead to the discovery of new biology, chemical tools, or even drug leads.

GB1 is a small (molecular weight < 500 g/mol) cyclodepsipeptide, unique from most cyanobacterial modified peptides or peptide–polyketide hybrids, which dominate the landscape of bioactive natural products produced by marine cyanobacteria (48). GB1 is densely functionalized, with all amino acids being modified and the presence of one hydroxy acid. Natural products possessing all these unusual structural features of GB1 (the hydroxamate, *C*-hydroxylated, and dehydroamino acids, and modified proline) have not been reported. The 4-MePro residue is a rare amino acid but has been previously found in cyanobacterial natural products (16). Interestingly, the hydroxamate group that is typical for metal chelators (especially iron siderophores) and present in other antiproliferative compounds (49) plays an iron-independent functional role in GB1’s binding to tubulin as the major mechanism of antiproliferative action. The additional metal binding ability potentially increases the pharmacological complexity of GB1 and remains to be investigated. However, the iron complexing ability

due to the hydroxamate functionality was proven but appeared not to play a predominant role with respect to the antiproliferative activity, based on the NCI-60 profile and cell cycle arrest characteristic for tubulin-targeting agents. We found the metal was necessary for neither GB1 binding to tubulin nor the inhibition of tubulin assembly into microtubules in our *in vitro* assays.

Tubulin-targeting agents strongly affect microtubule dynamics, which is a key feature for the functioning of these filaments in a wide range of biological processes. In fact, many of these compounds have been originally discovered within the extracts from living sources that often have mobility restrictions and, hence, use them as a protection mechanism against consumers. Some compounds promote microtubule stabilization through the interaction with either the taxane site (in the lumen of the microtubule) or the *pepo*/*lau* site (wall of the microtubule). The underlying mechanism is under discussion, but these compounds prevent microtubule depolymerization by locking lateral and/or longitudinal contacts. Microtubule-destabilizing compounds are more common and frequently more toxic. These induce rapid microtubule disassembly and prevent tubulin polymerization by two different molecular mechanisms: blockage (50, 51) and wedging (52, 53). Compounds that bind to the tip (*maytansine*) site (top surface of  $\beta$ -tubulin) or *pironetin* site (bottom surface of  $\alpha$ -tubulin) prevent the incorporation of a new tubulin heterodimer to the microtubule-growing end, due to the inactivation of the contact surfaces. Hence, these follow a blockage mechanism. Meanwhile, compounds interacting with the colchicine or the *vinca* site block the essential curve-to-straight conformational change in tubulin upon assembly, which necessarily comprise microtubule formation. Our GB1 high-resolution structure revealed a site and contributed to understanding the molecular mechanism underlying the compound’s destabilization effect. GB1 binds to a pocket localized between  $\alpha$ - and  $\beta$ -tubulin, close to the colchicine site and equivalent to the *vinca* site (that sits between two tubulin heterodimers), denoting a common underlying mechanism to destabilize microtubule formation: wedging. The colchicine site is an elongated pocket, where some compounds (colchicine, podophyllotoxin, or noscapine) bind to the first third, interacting with  $\beta$ -tubulin S9 and H7 (some also interact with the  $\alpha$ -tubulin T5 loop), whereas other compounds (e.g., nocodazole) bind deeper and also interact with the  $\beta$ -tubulin  $\beta$  sheet of the N-terminal domain (52, 54, 55). None have the extensive interactions with  $\alpha$ -tubulin that are observed with GB1.

Most tubulin-targeting agents bind to  $\beta$ -tubulin, which is the active molecule within the heterodimer upon guanosine-5′-triphosphate hydrolysis, with the exception of the *pironetin* site that is localized in  $\alpha$ -tubulin. Tubulin has been revealed as a complex molecule, since it includes two similar targeting sites that comprise interfaces of the two proteins, either within the heterodimer (*intradimer* interface, *gatorbulin* site) or between two heterodimers (*interdimer* interface, *vinca* site). Interestingly, these sites are equivalent, but contact surfaces are swapped between tubulin subunits. The top surface contains  $\alpha$ -tubulin (*vinca*) or  $\beta$ -tubulin (*gatorbulin*) residues, whereas the bottom surface contributes with  $\beta$ -tubulin (*vinca*) or  $\alpha$ -tubulin (*gatorbulin*) amino acids. Since tubulin is a key target for the treatment of cancer, this increases our options for the screening and development of new, safer, and more effective drugs. GB1 is a cyclic depsipeptide that represents a distinct chemotype that differs from other peptides targeting tubulin, such as dolastatin 10, and possesses low toxicity and molecular weight, adding to its promising small-molecule, drug-like properties and translational potential.

## Materials and Methods

All experimental details on the chemistry are provided in *SI Appendix*, including the isolation of the natural products, determination of the absolute configuration, and synthetic procedures. Biological methods and details about cell lines, proteins, and reagents are described in *SI Appendix*. Detailed descriptions are provided for the cell viability assessments, cell cycle analysis (HCT116 and HeLa cells), RNA isolation and RT-qPCR, indirect immunofluorescence,



caspase activity assays, tubulin polymerization, measurement of binding constants, tube formation assays, and migration and invasion assays. *SI Appendix, Table S1* contains NMR data for GB2, and *SI Appendix, Table S2* contains crystallography data collection and refinement statistics. NMR spectra of the natural products (GB1 and GB2) and all synthetic compounds are also provided as *SI Appendix, Figs. S1–S63*.

**Data Availability.** Atomic coordinates (crystallographic data) have been deposited in Protein Data Bank (PDB ID code 7ALR) (56).

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